

# भारत सरकार GOVERNMENT OF INDIA रक्षा मंत्रालय MINISTRY OF DEFENCE

# संयुक्त सेवा विनिर्देश JOINT SERVICES SPECIFICATION

ON

SODIUM NITRITE, TECHNICAL (DS Cat Part No. 6810-001 135) (NSN 6810720442726)

मानकीकरण निदेशालय रक्षा उत्पादन विभाग, रक्षा मंत्रालय छठी मंजिल 'ए' ब्लॉक रक्षा कार्यालय परिसर के जी मार्ग, नई दिल्ली-११०००१

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## RECORD OF AMENDMENTS

Amendment		Amendment pertains to:	Authority	Amended by	Signature
No.	Date	S. No./Para No./		Name & Appointment	&
		Column No.		(In Block Letters)	Date

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#### 0. FOREWORD

- **0.1** This specification has been prepared by Armament Standardisation Sub Committee on the authority of the Standardisation Committee, Ministry of Defence.
- **0.2** This specification has been approved and notified by the Director, Directorate of Standardisation, Ministry of Defence and is mandatory for use by the Defence Services.
- **0.3** The JSS 6810-104 : 2023 (Fifth Revision):
  - a) was prepared in the year 1993.
  - b) was revised in the year 2000, 2005, 2012 & 2017.
  - c) is revision of JSS 6810-104 : 2017 (Fourth Revision) and supersedes the same.
- **0.4** This specification would be used for Manufacture, Supply and Quality Assurance of Sodium Nitrite, Technical.
- **0.5** Quality Assurance Authority for the item covered in the JSS is the controller, Controllerate of Quality Assurance (Military Explosives), Aundh Road, Pune-411020 (email id cqamear-dgqa@nic.in). Enquiries regarding this specification, and drawings relating to any contractual conditions, shall be addressed to the quality assurance authority named in the tender or contract. Other enquiries shall be referred to:

The Director,
Directorate of Standardisation,
Ministry of Defence,
6<sup>th</sup> Floor, 'A' Block
Defence Offices Complex,
KG Marg, New Delhi-110001
Secretary ASSC, email id-assc.defstand@gov.in

**0.6** Non-registered users can obtain the following on payment:

## a) Copies of IS from:

Bureau of Indian Standards, Manak Bhawan, 9, Bahadur Shah Zafar Marg, New Delhi-110002

or

their regional/branch offices

## b) Copies of JSSs/JSGs from:

The Director,
Directorate of Standardisation
Standardisation Documents Centre,
Room No. 635, 6<sup>th</sup> Floor, 'A' Block,
Defence Offices Complex,
KG Marg, New Delhi-110001

**0.7** Indian Standard (IS) specifications are available free of cost for registered users on:

Directorate of Standardisation Website:

https://ddpdos.gov.in

For registration, visit our website

- **0.8** This specification holds good only for the supply order for which it is issued.
- **0.9 Directorate of Standardisation Website-**All the approved JSSs/JSGs are available on the Directorate of Standardisation Website *https://ddpdos.gov.in*. Defence Organisations desirous of accessing a copy of this document are requested to visit the Directorate of Standardisation website for registration and obtaining user id/password to access the JSSs/JSGs.

#### 1. SCOPE

This specification is meant to govern Manufacture, Supply and Quality Assurance of Sodium Nitrite, Technical for use in the manufacture of explosives.

#### 2. RELATED SPECIFICATIONS/DOCUMENTS

References are made in this specification to:

Table 1

S.	Specification/	Nomenclature
No.	Document No.	
a)	IS 138: 2018	Ready Mixed Paint, Marking, for Packages and
	(Fourth Revision)	Petrol Containers-Specification
b)	IS 323 : 2009	Rectified Spirit For Industrial Use-Specification
	(Second Revision)	
	Reaffirmed 2019	

#### 3. MATERIAL

Sodium Nitrite, Technical shall be in the form of clear crystals or lumps or sticks not more than pale yellow in colour and free from foreign matter, grit and visible impurities.

#### 4. MANUFACTURE

- **4.1** Sodium Nitrite, Technical shall be manufactured by a process which will produce the product conforming to this specification.
- **4.2** Nothing in this specification shall relieve the manufacturer of his responsibility for the safety of his operations during manufacture.

#### 5. TENDER SAMPLE

The manufacturer/supplier/contractor shall submit two tender sample of each 250 g essentially from the same batch/lot of the manufacture free of all charges and conforming to this specification, when called for in the tender to the Quality Assurance Officer/Quality Assurance Authority as stated in the tender.

### 6. PRE-INSPECTION OF STORES/CONSIGNMENT

6.1 Manufacturers/Contractors must satisfy themselves that the stores are in accordance with the terms of the contract and fully conform to the required specification by carrying out a thorough pre-inspection of each lot before actually tendering the same for inspection to the Quality Assurance Officer nominated under the terms of the contract. A declaration by the contractor that a necessary pre-inspection has been carried out on the stores tendered will be submitted along with the challan. The declaration will also indicate the method followed in

carrying out pre-inspection showing the features checked/tested and will have the test certificate attached to the challan/declaration.

**6.2** If the Quality Assurance Officer finds that the pre-inspection of the consignment as required above has not been carried out, the consignment is liable for rejection.

## 7. QUALITY ASSURANCE

### 7.1 Inspection

- **7.1.1** Sodium Nitrite, Technical and the packages in which it is packed shall be subjected to inspection by and to the approval of the Quality Assurance Officer/Quality Assurance Authority.
- **7.1.2** Samples of the material or the packages may be taken from any portion of the batch/lot/consignment.

## 7.2 Sampling

Two representative samples 250 g shall be drawn from each container. Normally the number of containers to be selected at random from a batch/lot shall depend on the size of the batch/lot and shall be in accordance with the following table:

No. of Containers in a Batch/Lot	No. of Containers to be Sampled
Up to 25	3
26 to 50	4
51 to 100	5
101 to 150	6
151 to 300	7
301 to 500	8
501 and above	10

## 7.3 Criteria for Conformity

- **7.3.1** If on examination, any sample is found not to conform to this specification, the whole batch/lot/consignment shall be rejected.
- **7.3.2** The foregoing provisions shall apply equally to prime contractors and to any subcontractor, if any.

#### 7.4 Test Requirements

Samples taken from any portion of batch/lot shall be in accordance with the clause 3 above and shall comply with the following test requirements:

Table 2

S.	Characteristics	Passing	Test Method
No.		Standard	
a)	Volatile matter, percent by mass	2.0 <i>Max</i>	Appx 'A'
b)	Sodium Nitrite (NaNO <sub>2</sub> ) content, percent	98.0 Min	Appx 'B'
	by mass		
c)	Matter insoluble in water, percent by	0.10 <i>Max</i>	Appx 'C'
	mass		
d)	Alkalinity (as Na <sub>2</sub> Co <sub>3</sub> ) percent by mass	0.20 <i>Max</i>	Appx 'D'
e)	Heavy Metals including Iron calculated	0.005 <i>Max</i>	Appx 'E'
	as Lead (Pb), percent by mass		
f)	Iron (as Fe <sub>2</sub> O <sub>3</sub> ), percent by mass	0.002 <i>Max</i>	Appx 'F'
g)	Sodium Nitrate, content (NaNO <sub>3</sub> ),	1.0 <i>Max</i>	Method as agreed
	percent by mass		between the
			purchaser and the
			contractor

**NOTE** - The percentage of tests from S. No. (b) to (g) shall be calculated on the dry mass of the material, free from moisture and volatile matter.

#### 8. WARRANTY

The stores supplied against this specification shall be deemed to bear warranty for 12 months from the date of receipt of store at consignee's end and against defective design/material/workmanship/performance. If during this period any of the stores supplied is found defective, the same shall be rectified/replaced by the contractor, free of charge, at the user's premises within a period of three months from date of intimation of defect.

#### 9. PACKAGING

- 9.1 Sodium Nitrite, Technical is deliquescent and shall be supplied in sound, clean and dry approved packages (25 kg in a heat-sealed polythene bag of 0.13 mm thickness placed inside a wooden drum or barrel).
- **9.2** Any other form of package shall have the prior approval of the Quality Assurance Officer/Quality Assurance Authority.
- **9.3** The inclusion of foreign matter or impurities in any of the package shall render the whole batch/lot/consignment liable to rejection.

## 10. MARKING

- **10.1** All packages containing the material shall be indelibly and legibly marked with the following details:
  - a) Nomenclature and Specification No. of the material.
  - b) Name and address of the Consignee.

- c) AT or SO No. and Date.
- d) Consignment No.
- e) Lot/Batch No. and date of manufacture.
- f) Gross and net mass.
- g) Consecutive No. of package and total number of packages in the consignment.
- h) Date of Supply.
- i) Manufacturer's/Contractor's initial or his recognised trademark.
- **10.2** In addition to the above the Quality Assurance Officer may suggest some more marking/identification suitable at the time of inspection.
- **10.3** The paint used for marking should conform to IS 138 and to the satisfaction of the Quality Assurance Officer/Quality Assurance Authority.

#### 11. DEFENCE STORES CATALOGUE NUMBER/NATO STOCK NUMBER

The Defence Stores Catalogue Number allotted to this store is 6810-001 135 and NATO Stock Number allotted to this stores is 6810720442726.

#### 12. SAFETY OF OPERATION

Nothing in this specification shall relieve the manufacturer/supplier/contractor/user of his responsibility for the safety of operations in the manufacture, storage, transit or use of this store.

#### 13. SUGGESTIONS FOR IMPROVEMENT

Any suggestion for improvement in this document may be forwarded to:

The Director,
Directorate of Standardisation,
Ministry of Defence,
6<sup>th</sup> Floor, 'A' Block,
Defence Offices Complex
KG Marg, New Delhi-110001

## APPX 'A'

(Clause 7.4, Table 2, S. No. (a))

## **DETERMINATION OF VOLATILE MATTER**

**A-1.** Weigh accurately by difference about 2 g of the material in a squat form stoppered glass weighing bottle  $(M_2)$  and dry to constant mass in a vacuum desiccator over freshly regenerated silica gel  $(M_1)$ . Express the loss in mass as the percentage of the mass of the material taken for test.

## A-2. CALCULATIONS

where,

 $M_2$  = Mass of the weighing bottle + sample (before desiccation); and

 $M_3$  = Mass of the weighing bottle + sample (after desiccation).

#### APPX 'B'

(Clause 7.4, Table 2, S. No. (b))

#### DETERMINATION OF SODIUM NITRITE CONTENT

#### **B-1. PREPARATION OF SAMPLES SOLUTION**

Dissolve about 1.2 g of the dried material, accurately weighed, in water and dilute it to exactly 100 ml in a volumetric flask.

#### **B-2. REAGENTS**

The following reagents are required:

- 1) Concentrated Sulphuric acid Specific Gravity 1.84.
- 2) Standard Potassium Permanganate Solution Approximately 0.1 N but of known normality  $(N_1)$ .
- 3) Standard Ferrous Ammonium Sulphate solution Approximately 0.1 N but of known normality  $(N_2)$ .

#### **B-3. PROCEDURE**

Take 300 ml of freshly boiled and cooled distilled water and 5 ml of concentrated Sulphuric acid in an Erlenmayer flask and then immediately add standard Potassium Permanganate solution until a faint colour persists for 2 minutes. Disregard this volume of the Permanganate solution. Add from a pipette, exactly 50 ml of standard Potassium Permanganate solution clause **B-2(2)** mix thoroughly and slowly add with stirring 10 ml of the prepared sample solution clause **B-1**, care being taken to hold the tip of the pipette under the surface of the liquid during the addition. Warm the mixture of 40°C and maintain at this temperature of 10 minutes. Cool the mixture to 10°C and then add 15 ml of the standard Ferrous Ammonium Sulphate solution clause **B-2(3)**. (Colour of Permanganate should be discharged by this volume of Ferrous Ammonium Sulphate solution should be added). Let it stand for 5 minutes and titrate the excess of Ferrous Ammonium Sulphate with standard Potassium Permanganate solution.

#### **B-4. CALCULATION**

where,

X = Volume of Potassium Permanganate solution required to back titrate the excess of Ferrous Ammonium Sulphate solution;

$N_1$	=	Normality of standard Potassium Permanganate solution;
V	=	Volume in ml of Ferrous Ammonium Sulphate solution required for excess Potassium permanganate solution;
$N_2$	=	Normality of standard Ferrous Ammonium Sulphate solution; and
M	=	Mass in g of the dried material taken from the prepared sample solution

#### APPX 'C'

(Clause 7.4, Table 2, S. No. (c))

#### DETERMINATION OF MATTER INSOLUBLE IN WATER

- **C-1.** Dissolve about 20 g of the material dried and accurately weighed, in 150 ml of water in a 300 ml beaker and heat on a steam bath for one hour. Filter through a tared Gooch crucible previously washed and dried at 105°C to 110°C. Cool and weigh. Repeat the drying till constant mass is obtained.
- **C-2.** Determine the mass of the insoluble residue and express the result as percentage of the mass of the dried material taken for test.

## C-3. CALCULATIONS

Total matter insoluble in water Percent by mass = Mass of residue x 100

-----Mass of sample taken

APPX 'D'

(Clause 7.4, Table 2, S. No. (d))

#### **DETERMINATION OF ALKALINITY**

#### D-1. REAGENTS

The following reagents are required:

- 1) Methyl Red Indicator Dissolve 0.1 g of the Methyl red in 200 ml of rectified spirit, 95% by volume (conforming to IS 323).
- 2) Standard Hydrochloric Acid Approximately 0.1 N.

#### D-2. PROCEDURE

Dissolve 5 g of the material, dried and accurately weighed, in 100 ml to 200 ml of water. Add 2 drops of Methyl red indicator and titrate with standard Hydrochloric acid.

#### D-3. CALCULATIONS

Alkalinity (as 
$$Na_2CO_3$$
), percent by mass = 
$$\frac{5.3 \text{ VN}}{M}$$

where,

V = Volume in ml of standard Hydrochloric acid used for the test

N = Normality of standard Hydrochloric acid and

M = Mass in g of the dried material taken for the test

#### APPX 'E'

(Clause 7.4, Table 2, S. No. (e))

#### DETERMINATION OF HEAVY METALS INCLUDING IRON

#### E-1. APPARATUS

#### E-1.1 Nesseler Tubes

Two flat bottomed tubes of thin colourless glass, about 25 mm in diameter and about 150 mm in length, graduated at 50 ml. The depth, measured internally from the graduation mark to the bottom shall not vary by more than 2 mm in the two tubes.

#### E-2. REAGENTS

The following reagents are required:

- 1) Concentrated Hydrochloric acid Specific gravity 1.16.
- 2) Standard Lead solution Dissolve 1.60 g of Lead Nitrate in water, add 1 ml of concentrated Nitric acid and make the volume upto 1000 ml. Transfer exactly 10 ml of this solution to a volumetric flask, again dilute with water and make up volume to 1000 ml mark. One millilitre of this solution is equivalent to 0.01 mg of Lead (Pb).
- 3) Acetic acid Approximately 1 N.
- 4) Hydrogen Sulphide solution Prepare a fresh, saturated aqueous solution of Hydrogen sulphide gas.

#### E-3. PROCEDURE

- **E-3.1** Weigh 10 g of the material, dried as described in Appx 'A' above and dissolve in 50 ml of water. Add 10 ml of concentrated Hydrochloric acid and evaporate to dryness on the water bath. Dissolve the residue in water and make up the solution to exactly 500 ml (solution X).
- **E-3.2** Take 10 ml of solution X in a Nessler tube and add 2 ml of standard Lead solution. Dilute to 30 ml and add 1 ml of Acetic acid (Solution Y).
- **E-3.3** Take 30 ml of solution X in another Nessler tube and add 1 ml of Acetic acid (Solution Z).
- **E-3.4** Add 10 ml of Hydrogen sulphide solution to each of solution Y and Z. The limit prescribed shall be taken as not having been exceeded if the intensity of colour produced in solution Z is not greater than that of Solution Y.

#### APPX 'F'

(Clause 7.4, Table 2, S. No. (f))

#### **DETERMINATION OF IRON**

#### F-1. APPARATUS

#### F-1.1 Nesserler Tubes

Same as under Appx 'E' Clause E-1.1

#### F-2. REAGENTS

The following reagents are required:

- 1) Concentrated Hydrochloric Acid Specific Gravity 1.16.
- 2) Butanolic Potassium Thiocyanate Solution Dissolve 10 g of Potassium Thiocyanate in 10 ml of water of  $27^{\circ}\text{C} \pm 2^{\circ}\text{C}$ . Add sufficient butanol to make to 100 ml and shake vigorously until the solution is clear.
- 3) Standard Iron Solution Dissolve 0.6040~g of Ferric Ammonium Sulphate in 10~ml of dilute Sulphuric acid (10% by volume) and dilute with water to 1000~ml. Take 10~ml to this solution and dilute to 100~ml. One millilitre of the final solution is equivalent of 0.01~mg of Ferric oxide ( $Fe_2O_3$ ).

## F-3. PROCEDURE

Dissolve 1 g of the dried material in 10 ml of water, add 1 ml of concentrated Hydrochloric acid and 15 ml of Butanolic Potassium Thiocyanate Solution, shake vigorously for thirty seconds and allow to separate. Carry out a control test in the other Nesserler Tube using 2 ml of standard Iron solution in place of the material and the same quantities of other reagents in the same total volume of the reaction mixture. Compare the colour produced in the two tubes after 5 minutes.

**F-4.** The limit prescribed shall be taken as not having been exceeded if the intensity of the colour produced in the test with the material is not greater than that produced in the control.